Latent Heath and Unit 7 Exam

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| ***Teacher:*** | Ms. Athwal | ***Date:*** | April 28 – May 2 | ***Course:*** | Chemistry | ***Grade:*** | 11 |
| ***CA Standard(s):***  8a *Students know* the rate of reaction is the decrease in concentration of reactants or the increase in concentration of products with time. | | | | | | | |
| ***Learning Objective (s):***  8.1 – I can describe the rate of a reaction using the ideas of concentration of reactants, concentration of products, and time. I can explain why rate is dependent on concentration. | | | | | | | |
| * ***Essential Question(s):*** What is McDonald’s Putting In Its French Fries? –Why don’t they ever go bad? | | | | | | | |
| **Assessment**:   * Unit 7 Exam Test Corrections * Equilibrium POGIL Packet * Factors on rate of reaction exit ticket | | | | | | | |
| ***Do Now***:   1. Review Vocabulary | | | | | | | |

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| **WHOLE GROUP/ DIRECT INTRUCTION** |
| * POGIL Equilibrium Packets * Factors affecting reactions rates lescture 8.1 |

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| **SMALL GROUP STATION** |  | **COLLABORATIVE STATION** |  | **COMPUTER ASSISTED STATION** |
| Notes on factors that influence reaction rates (8.1.Notes on Google Drive) |  | [Kinetics and Equilibrium POGIL](../../../../Chemistry%20Tool%20Kit/POGIL/31%20Equilibrium-S.pdf) |  |  |